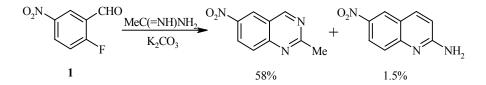
## CYCLOCONDENSATION OF 2-FLUORO-5-NITROBENZALDEHYDE WITH AMIDINES

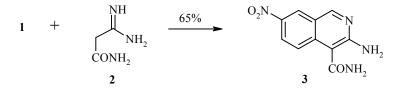
## D. V. Dar'in, S. I. Selivanov, P. S. Lobanov, and A. A. Potekhin

Keywords: amidines as N,N- and C,N-nucleophiles, 2-fluoro-5-nitrobenzaldehyde, cyclocondensation.

Kotsuki et al. [1] have recently reported the synthesis of 4-unsubstituted quinazolines by the cyclocondensation of *o*-fluorobenzaldehydes with amidines. Waring [2] and Mohamed [3] have reported that amidines containing  $\alpha$ -hydrogen atoms may undergo cyclocondensation with esters of 3-oxocarboxylic acids and conjugated enols by acting not only as N,N-dinucleophiles but also as C,N-dinucleophiles. These findings have led us to study the reaction of 2-fluoro-5-nitrobenzaldehyde (1) with acetamidine under the same conditions as in the work of Kotsuki [1] and thoroughly separated the reaction mixture by column chromatography. The major reaction product, as expected, was 2-methyl-6-nitroquinazoline. Furthermore, isomeric 2-amino-6-nitroquinone formed in the nucleophilic attack of the amidine  $\alpha$ -carbon atom on the benzaldehyde carbonyl group was obtained in trace amounts. Examples of cyclocondensation featuring unsubstituted acetamidine as a C,N-dinucleophile have not yet been reported.



When amidine **2**, containing an active methyl group, was introduced into the same reaction, 3-aminoisoquinoline **3** was the major cyclocondensation product.



Isoquinoline structure was assigned for **3** on the basis of the chemical shifts for 1-H and  $C_{(1)}$  and the  ${}^{1}J(C^{1}-H^{1})$  coupling constant. These values are characteristic for a C–H fragment in the  $\alpha$ -position to the nitrogen atom in azines [5].

**2-Methyl-6-nitroquinazoline** was isolated in 58% yield by chromatography on a silica gel column with 1:2 ether–hexane as the eluent; mp 152-154°C. <sup>1</sup>H NMR spectrum (300 MHz, DMSO-d<sub>6</sub>),  $\delta$ , ppm, *J* (Hz): 2.84 (3H, s, CH<sub>3</sub>); 8.05 (1H, d, *J* = 9, 8-H); 8.61 (1H, m, *J* = 9, 2-, 5-, 7-H); 9.12 (1H, d, *J* = 5-H); 9.78 (1H, s, 4-H). Found, %: C 57.21; H 3.72; N 22.10. C<sub>9</sub>H<sub>7</sub>N<sub>3</sub>O<sub>2</sub>. Calculated, %: C 57.14; H 3.73; N 22.21.

1014

St. Petersburg State University, 198904 St. Petersburg, Russia; e-mail: psl@pisem.net. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 8, pp. 1155-1156, August, 2002. Original article submitted April 24, 2002.

**2-Amino-6-nitroquinoline** [14] was isolated in 1.5% yield by chromatography on a silica gel column with 3:1:1 ether–hexane–ethyl acetate as the eluent; mp 250°C (dec.). <sup>1</sup>H NMR spectrum,  $\delta$ , ppm, *J* (Hz): 6.89 (1H, d, *J* = 9, 3-H); 7.23 (2H, NH<sub>2</sub>); 7.50 (1H, d, *J* = 9, 8-H); 8.13 (1H, d, *J* = 9, 4-H); 8.22 (1H, m, *J* = 9, 2-, 5-, 7-H); 8.68 (1H, d, *J* = 2.5, 5-H).

**3-Amino-7-nitroisoquinoline-4-carboxamide (3)** was isolated in 65% yield by reprecipitation from DMF using 1:1 ether–ethanol; mp ~350°C. <sup>1</sup>H NMR spectrum,  $\delta$ , ppm, *J* (Hz): 6.86 (2-H, 3-NH<sub>2</sub>); 7.83 (2H, d, *J* = 9, 5-H); 7.85 (1H, CONH); 8.06 (1H, CONH); 8.24 (1H, m, *J* = 9, 2-, 5-, 6-H); 8.93 (1H, d, *J* = 2.5, 8-H); 9.19 (1H, s, 1-H). <sup>13</sup>C NMR spectrum,  $\delta$ , ppm: 104.35 (C<sub>(4)</sub>), 119.79 (C<sub>(8a)</sub>), 123.83 (C<sub>(6)</sub>), 126.14 (C<sub>(8)</sub>), 137.41 (C<sub>(4a)</sub>), 140.98 (C<sub>(7)</sub>), 155.98 (C<sub>(3)</sub>), 156.12 (C<sub>(1)</sub>, <sup>1</sup>*J*<sub>(C-H)</sub> = 181.5 Hz), 168.08 (CO). Found, %: C 51.66; H 3.52; N 24.19. C<sub>10</sub>H<sub>8</sub>H<sub>4</sub>O<sub>3</sub>. Calculated, %: C 51.73; H 3.47; N 24.13.

## REFERENCES

- 1. H. Kotsuki, H. Sakai, H. Morimoto, and H. Suenaga, *Synlett*, 1993 (1999).
- 2. D. J. Brown and P. Waring, Austral. J. Chem., 30, 621 (1977).
- 3. M. A.-M. Gomaa, A. M. Nour El-Din, and A. A. Mohamed, Bull. Chem. Soc. Jpn., 72, 471 (1999).
- 4. A. E. Tschitschibabin, D. P. Witkovsky, and M. I. Lapschin, Ber., 58, 803 (1925).
- 5. H.-O. Kalinowsky, S. Berger, and S. Braun, <sup>13</sup>C-NMR Spektroskopie, Georg Thieme Verlag, Stuttgart– New York (1984), pp. 356, 455.